GOVERNMENT COLLEGE (A), RAJAMAHENDRAVARAM I B.Sc. ANALYTICAL CHEMISTRY SEMESTER – II QUESTION PAPER BLUE PRINT Course -4: BASIC PRINCIPLES IN ANALYTICAL CHEMISTRY

TIME: $2^{1/2}$ hrs.

MARKS: 50 M

PART -A

Answer ALL the Questions

1. 2 Questions from UNIT- I

2. 2 Questions from UNIT- II

3. 2 Questions from UNIT-III

4. 2 Questions from UNIT IV

5. 2 Questions from UNIT-V

PART - B

Answer any FIVE Questions

6. 1 Question from UNIT- I

7. 2 Questions from UNIT- II

8. 2 Question from UNIT- III

9. 2 Questions from UNIT- IV

10. 1 Questions from UNIT- V

5 x 7 = 35 M

5 x 3 = 15 M

GOVERNMENT COLLEGE (A), RAJAHMUNDRY I B.Sc. ANALYTICAL CHEMISTRY SEMESTER-II MODEL PAPER (From 2023-234) Course - 4: BASIC PRINCIPLES IN ANALYTICAL CHEMISTRY

Time: $2^{1/2}$ hrs.

PART- A

Answer ALL the questions. Each carries SEVEN marks

 $5 \mathbf{x} \mathbf{7} = \mathbf{35} \mathbf{M}$

Maximum Marks: 50

1. What are Primary standard solutions? Explain the experimental procedure for the preparation of a primary standard solution.

(**OR**)

- 2. What are secondary standard solutions? Explain the experimental procedure for the preparation of a secondary standard solution.
- 3. Explain the methods of detecting analytes based on Electic charge and electromagnetic radiation

(OR)

- 4. Explain about the Instrumental methods of analysis with examples
- 5. Explain in detail about pH meter

(OR)

- 6. Write in detail about the single pan analytical balance
- 7. What are Errors? Write about the classification of errors

(**OR**)

- 8. What are significant figures? Write the computation rules of significant figures
- 9. Explain different theories of Indicators

(**OR**)

10. Explain the acid – base titrations with examples

PART- B

Answer any FIVE of the following questions. Each carries THREE marks 5 x 3 = 15 M

- 11. Define mole and molar concentration
- 12. Write briefly about chemical analysis
- 13. Discuss Qualitative analysis in briefly
- 14. How do you calibrate volumetric flask?
- 15. Write the uses of commonly used laboratory glass apparatus
- 16. Define mean and standard deviation
- 17. Explain briefly about accuracy
- 18. What are Buffer solutions?